

Code No: **R42048**

**R10**

**Set No. 1**

**IV B.Tech II Semester Regular/Supplementary Examinations, April - 2015**

**ANALYTICAL INSTRUMENTATION**

**(Electronics and Communication Engineering)**

**Time: 3 hours**

**Max. Marks: 75**

**Answer any FIVE Questions  
All Questions carry equal marks**

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- 1 a) Explain the importance of pH measurement in process industries and explain the process of pH measurement. [8]  
b) Explain the conductivity method for the measurement of SO<sub>2</sub> in air sample. [7]
- 2 a) State and explain Beer lambert law. Explain the deviations of this law [8]  
b) Describe in detail, how galvanic method can be used to determine the Oxygen concentration in a gas. [7]
- 3 a) Explain in detail Electron capture detector used in the process of gas chromatography system with neat sketches. [8]  
b) Explain in carrier flow diagram and the operation of liquid chromatography with neat sketches. [7]
- 4 a) Write about the paramagnetic type detectors [8]  
b) Explain in detail sample injection system in chromatography. [7]
- 5 a) What are the various detectors normally used in UV/VIS spectrometers? Describe the construction and working of any one of them [8]  
b) Explain the advantages and disadvantages of single and double beam spectrometer? [7]
- 6 a) How is grating created? What is their specific feature and where is it used? [8]  
b) Briefly discuss about the detectors used for flame photometry. [7]
- 7 a) Explain the constructional details of NMR spectrometer [8]  
b) Brief discussion on electron spin resonance [7]
- 8 a) Describe the properties of  $\alpha$ ,  $\beta$  and  $\gamma$  particles in brief. [8]  
b) What are Counters? What for and why are these used in nuclear radiation measurements? [7]

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**Set No. 2**

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**(Electronics and Communication Engineering)**

**Time: 3 hours**

**Max. Marks: 75**

**Answer any FIVE Questions**  
**All Questions carry equal marks**

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- 1 a) With schematic diagram explain the Chopper amplifier type pH meter. [8]  
b) Explain how the output potential of pH electrode is related to pH and also for  $H^+$  ion concentration with equations. [7]
- 2 a) Explain the principle of operation of thermal conductivity analyser used for estimation of components such as  $CO$ ,  $CO_2$  etc. [8]  
b) With a neat block diagram explain any one type of sodium analyser. [7]
- 3 a) Explain in detail the sample injection system in chromatography. [8]  
b) Explain the operation of multichannel type instrument to calculate focal length of a monochromator. [7]
- 4 a) Sketch the components of gas chromatography. [8]  
b) Explain in detail photo Ionisation detector used in the process of gas chromatography. [7]
- 5 a) Discuss about the simple analytical procedures on given samples using Ultraviolet Spectrophotometers. [8]  
b) Explain in detail the optical filters used in spectrometry. [7]
- 6 a) Describe the basic difference between a spectrometer for absorption measurements and emission studies. [8]  
b) Briefly discuss about the detectors of flame photometry. [7]
- 7 a) Discuss the merits of various types of mass spectrometer. [8]  
b) Draw the schematic and explain the multi-channel spectrophotometer [7]
- 8 a) Explain the factors affecting the counting of pulses. [8]  
b) Discuss electromagnetic radiation, its properties and its interaction with matter in brief. [7]

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**Set No. 3**

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**ANALYTICAL INSTRUMENTATION**

**(Electronics and Communication Engineering)**

**Time: 3 hours**

**Max. Marks: 75**

**Answer any FIVE Questions**

**All Questions carry equal marks**

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- 1 a) With schematic diagram explain the vibrating condenser amplifier pH meter. [8]  
b) Explain how the thermal conductivity of hydrogen different from other gas analysers. [7]
- 2 a) How can IR gas analyser be used to determine the components such as CO, CO<sub>2</sub> and Nitric oxide of a given sample. [8]  
b) Explain the construction details of sodium analyser. [7]
- 3 a) What are the various applications of gas chromatography and explain any one application in detail [8]  
b) Describe in detail, how galvanic method can be used to determine oxygen concentration of a gas [7]
- 4 a) Explain the paramagnetic type detectors used in gas chromatography. [8]  
b) Discuss the simple analytical procedures on given samples using Gas chromatographic methods. [7]
- 5 a) Differentiate photodiode array detector and diode array detector [8]  
b) What is Non dispersive absorption type IR Technique-briefly explain. [7]
- 6 a) Explain the sources for Flame Photometers. [8]  
b) By way of a schematic explain the operation of non-dispersive dual channel absorption technique of IR spectrometer [7]
- 7 a) Explain briefly about the atomic emission and absorption spectroscopy [8]  
b) The order used in an echelle grating is 60 and the dispersion angle of 620, groove density 84/mm and the focal length of 0.6 m, obtain the reciprocal linear dispersion and resolution. [7]
- 8 a) What are the possible radiation methods with different interaction techniques? [8]  
b) Explain the Industrial application of radiation measurement. [7]

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**Set No. 4**

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**ANALYTICAL INSTRUMENTATION**  
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**Time: 3 hours**

**Max. Marks: 75**

**Answer any FIVE Questions**  
**All Questions carry equal marks**

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- 1 a) Explain how sodium is analysed in boiler water of a power plant. [8]  
b) Explain the hydrogen purity meter by using thermal conductivity method. [7]
- 2 a) Explain the principle of paramagnetic oxygen analyser [8]  
b) Write about the estimation of sulphur dioxide using conductivitometry. [7]
- 3 a) What are the analytical procedures on given samples using Liquid chromatographic methods? [8]  
b) Discuss the applications in which gas chromatography is particularly useful. [7]
- 4 a) Explain the electron capture detector used in the process of gas chromatography. [8]  
b) Explain in detail photo Ionisation detector used in the process of gas chromatography system with neat sketches. [7]
- 5 a) Explain about the merits and demerits associated with single and double beam spectrometer? [8]  
b) How does IR spectrophotometer analyze various substances? Discuss. [7]
- 6 a) How is online calorific value measurements made? Explain. [8]  
b) Enumerate some of the major applications of UV photometry in brief. [7]
- 7 a) Explain the constructional details of NMR spectrometer [8]  
b) Write about the principle on electron spin resonance. [7]
- 8 a) Discuss the Industrial application of Gamma Spectrometry. [8]  
b) Draw the characteristics of radiation counter tube. [7]